Lesson 4. Stochiometric air requirement calculation Question Bank

1. The molecular weight of carbon dioxide is
2 kg of oxygen is required to burn one kg of methane.
3. The amount of oxygen in air is% by weight
4. If the fuel has one kg of sulphur, on complete combustion it produces $___$ kg of SO_2
5. More amount of oxygen in the combustion product indicatecombustion
6. Give the step by step procedure for the determination of stochiometric air requirement.
7. How to assess the products of combustion of a fuel.